# **Application of** Rarefied Flow & Plasma Simulation Software







Yokohama City in Japan

WAVE FRONT

## Profile of Wave Front Co., Ltd.

Name: Wave Front Co., Ltd.

Incorporation: March 1990

Head Office: Yokohama in Japan



#### **Business Description**

- (1) CFD(Computational Fluid Dynamics)&Plasma Software, Sales & Consulting Services
- (2) CMMS(Computerized Maintenance Management System) Software, Sales, Consulting and Implementation services

### Software Products sold in Foreign Countries

- (1) Particle-PLUS( Plasma) & DSMC-Neutrals( Rarefied Gas Flow)
- (2) PM-Optimizer( CMMS) & FLIPS( Scheduler)





# Rarefied Neutral Gas Flow Simulation using Particle Method

## Feature

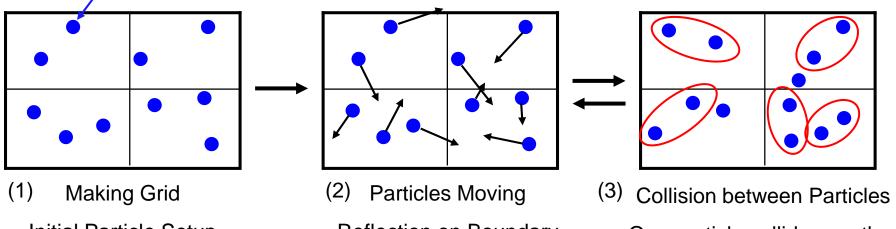


- Rarefied Neutral Gas Flow Simulation Software based on Direct Simulation Monte Carlo (DSMC) method
- 2. Simulation of Thin Film Growth due to Chemical Reaction like CVD
  - Gas Phase Chemical Reaction
  - Surface Chemical Reaction
- 3. Automatic and Very Fast Mesh Generation
- 4. Unstructured Mesh for Modeling Detail Geometry
- High Performance of Parallel Computing and Applicability to Big Model
- 6. No Divergence for All Model including Bad Quality Mesh User can obtain solution always.

## DSMC method







Initial Particle Setup

- Reflection on Boundary
- One particle collides another in the same cell.

Requirement of numerical parameters

Cell size

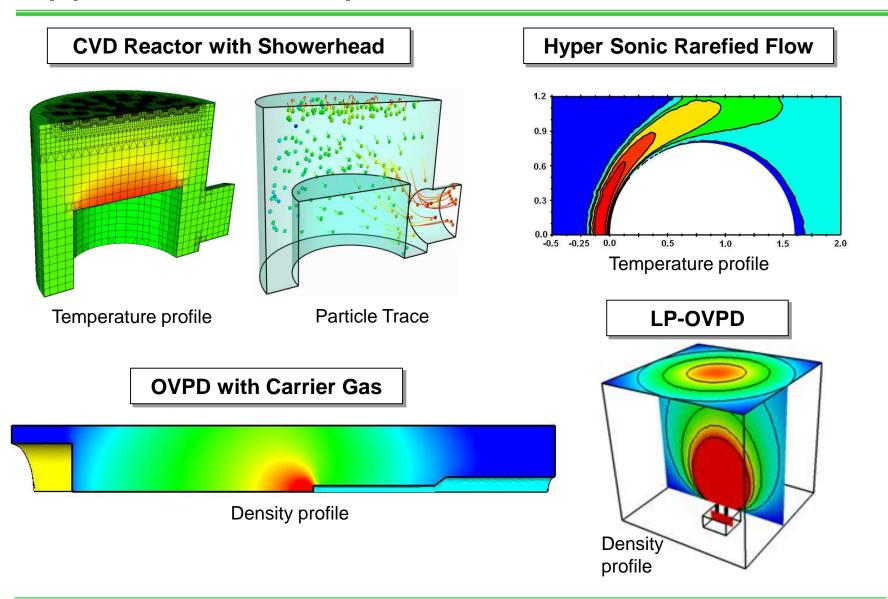
 $\Delta x < \lambda$ : mean free path

Time interval

 $\Delta t < \tau$ : collision time

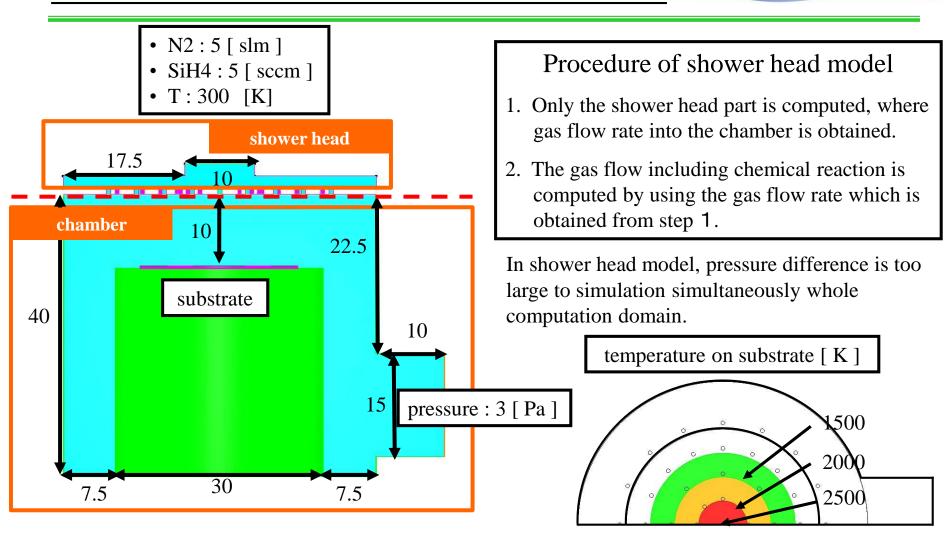
# Application examples





## 1. CVD model with Shower head





\* Note that this model is just validation model in order to deposition rate depending on temperature. Therefore, computation setting is not realistic.

## 1. Chemical reaction of CVD model



$$k_f = \Lambda T^{\eta} \exp\left(-\frac{E_a}{T}\right)$$

### Gas phase chemical reaction\*1

Reaction equation	$\Lambda[m3/s]$	η	$E_a[K]$
SiH4 + M -> SiH2 + H2 + M	0.865	-3.54	29000

<sup>\*1</sup> J. Phys. Chem. vol.98 10138 (1994)

#### Surface reaction\*2

Surface reaction	$\Lambda[m3/s]$	η	$E_a[K]$
SiH2 -> Si(s) + H2	1	0	0
SiH4 -> Si(s) + 2H2	0.054	0.0	9400

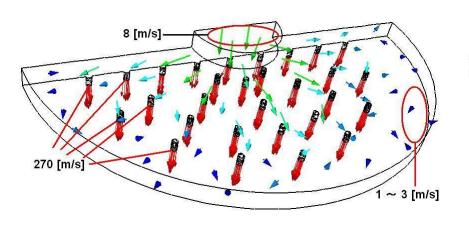
<sup>\*2</sup> J. Crystal Growth vol.126 373 (1993)

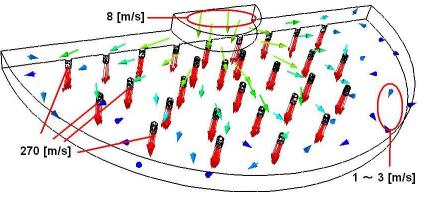
# 1. Shower head part



N2 flow velocity [ m/s ]

SiH4 flow velocity [ m/s ]

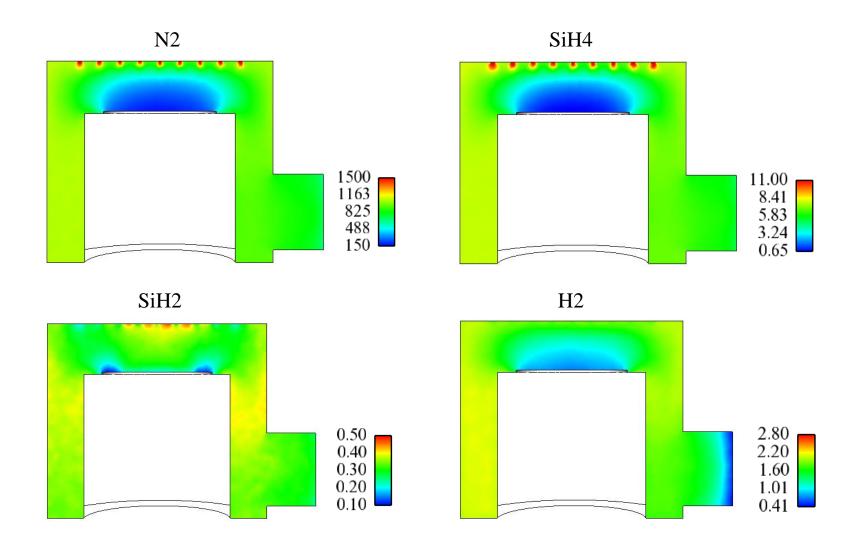




- ➤ Both flow velocity profiles is almost the same because pressure is sufficiently high.
- Solution Gas flow rates of each shower head nozzle is obtained. The gas flow in the chamber using them.

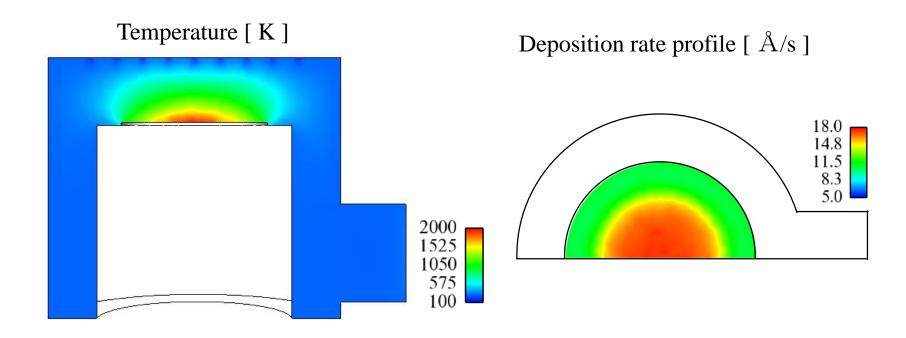
# 1. Number density [10<sup>18</sup> #/m<sup>3</sup>]





# 1. Temperature and depo. rate

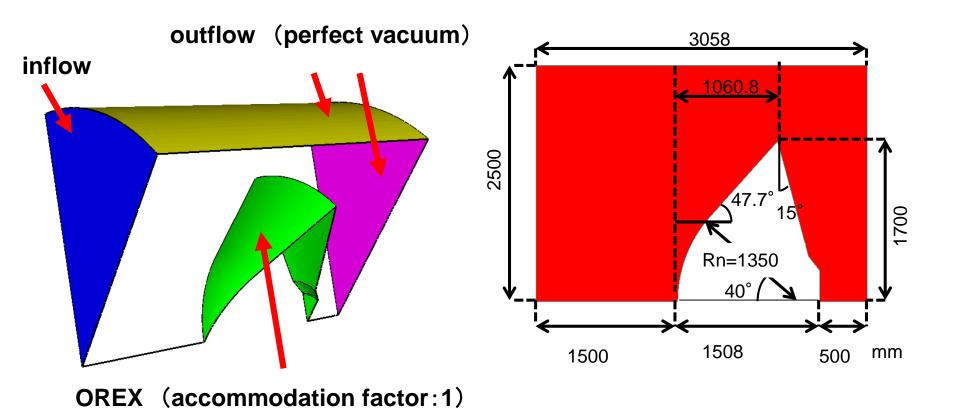




- > Temperature jump is reproduce on substrate.
- The temperature dependence of deposition rate appears on the substrate.

# 2. Hyper Sonic Rarefied Flow





altitude [km]	flow vel. [m/s]	density [m <sup>-3</sup> ]	mole frac. (O2)	mole frac. (N2)	mole frac. (O)
105.0	7451.0	5.0515E+18	0.1528	0.7815	0.0657
92.8	7454.1	4.0845E+19	0.2025	0.7881	0.0094

## 2. Chemical reaction



Gas phase chemical reaction is computed using TCE model. The molecules used in the computation consider vibrational and rotational energy.

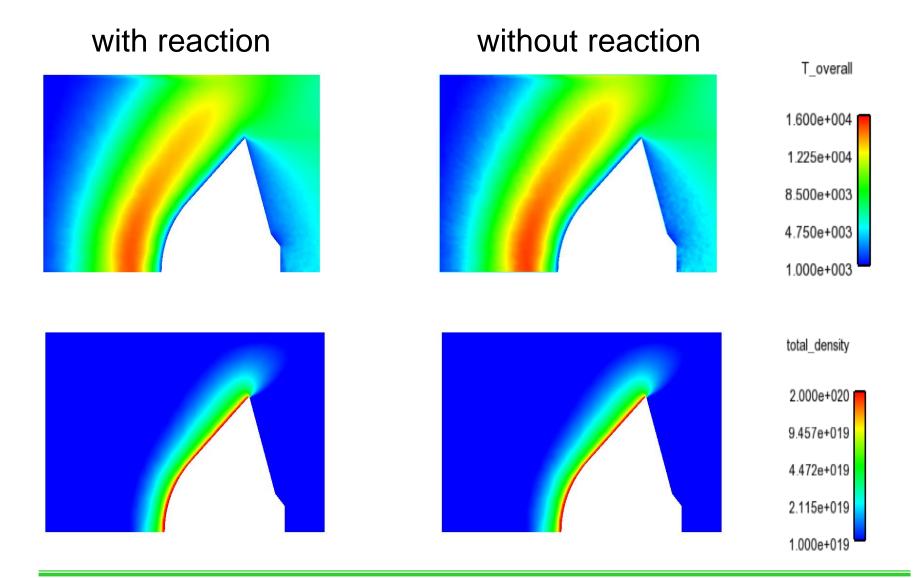
Reaction equation	A [cm³/mol s]	В	Ea [K]
N2 + M <-> N + N + M	7.0x10 <sup>21</sup>	-1.6	113200
O2 + M <-> O + O + M	2.0x10 <sup>21</sup>	-1.5	59500
NO + M <-> N + O +M	5.0x10 <sup>15</sup>	0.0	75500
NO + O <-> N + O2	8.4x10 <sup>12</sup>	0.0	19450
N2 + O <-> N + NO	6.4x10 <sup>17</sup>	-1.0	38370

$$k_f = A T^B \exp\left(-\frac{E_a}{T}\right)$$

<sup>\*3</sup> Bird G. A., *Molecular Gas Dynamics and the Direct Simulation of Gas Flows*, 1st edition, Clarendon Press, Oxford, New York, 1994.

# 2. Results (altitude: 105 km)





# 2. Heat flux [kW/m<sup>2</sup>]



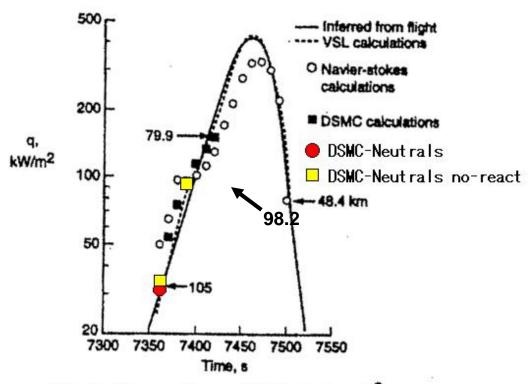


Fig 3. Comparison of flight inferred<sup>3</sup> stagnation-point heating rates with calculations using VSL<sup>7</sup>, NS<sup>6</sup>, and DSMC.

<sup>\*</sup> DSMC SIMULATION OF OREX ENTRY CONDITIONS: James N. Moss, Roop N. Gupta, Joseph M. Price (Aerothermodynamics Branch, NASA Langley Research Center, Hampton, Virginia 23681-0001, USA)

# 2. Heating rate coefficient



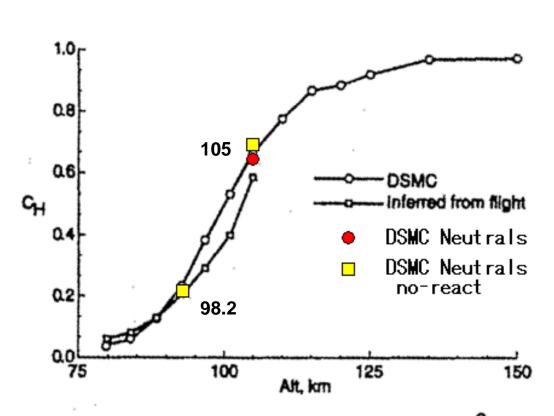


Fig. 4. Comparison of flight inferred<sup>3</sup> stagnation-point heating rate coefficient with DSMC results.

$$C_{\rm H} = \frac{2q}{\rho V^3}$$

mass density in atmospheric pressure

$$\rho = \frac{\sum_{s} M_{s} n_{s}}{N_{A}}$$

q : heat flux

M<sub>s</sub>: atomic mass

n<sub>s</sub>: No. density

 $N_A$ : Avogadro constant

V : OREX's velocity



# Low Pressure Plasma Simulation using Particle Method

## Feature of Particle-PLUS



- 1. Plasma Simulation Software Package for Law Pressure
- 2. Composition of Main Two Modules,

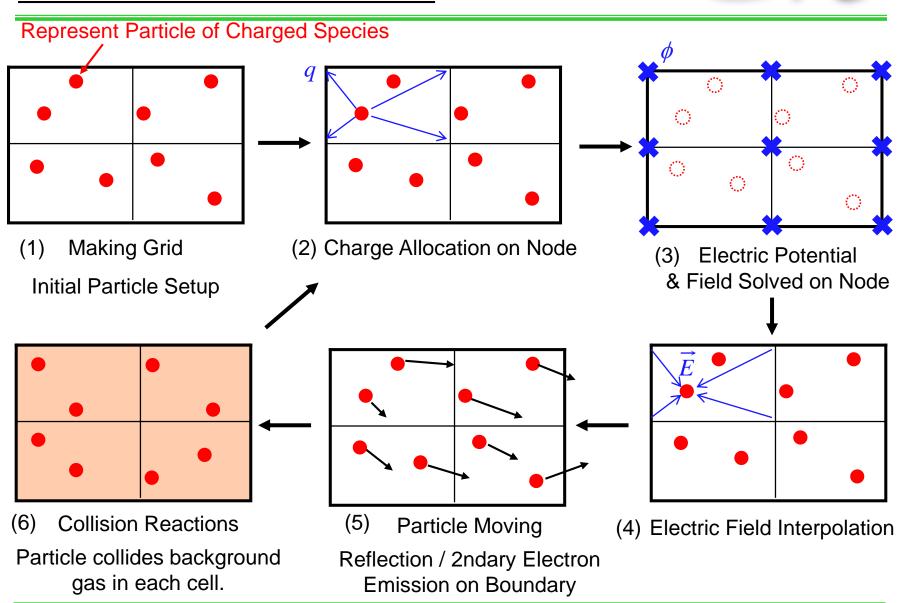
Plasma Module: based on Particle-in-Cell (PIC) method

Neutral Module: based on DSMC method

- 3. Simulation of Thin Film Growth due to Magnetron Sputtering
  - Background Gas Flow: Neutral Module
  - Magnetic Field by External Magnet: Plasma Module
  - Collision Reactions by Electron/Ion Impact: Plasma Module
  - Secondary Electron Emission : Plasma Module
  - Sputtering of Target Atom: Neutral Module
  - Motion of Sputtered Atom: Neutral Module

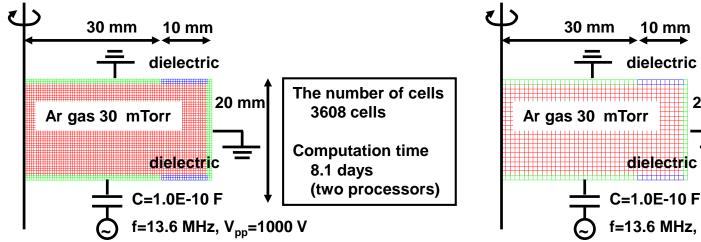
## Particle in cell method

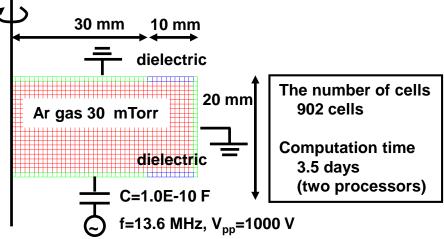


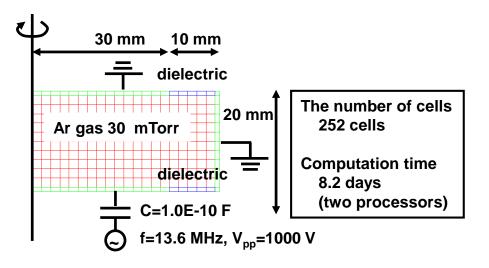


## Mesh size dependence



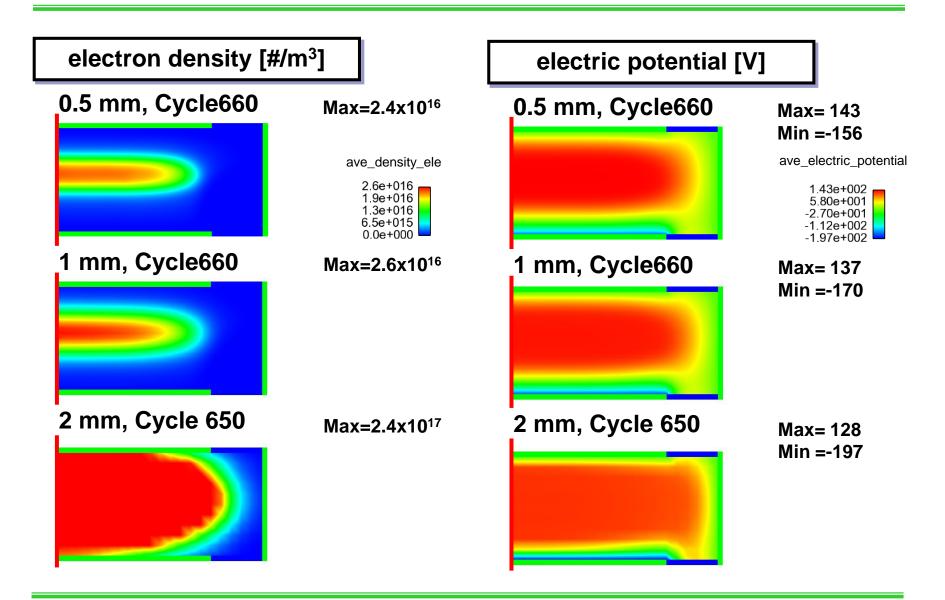






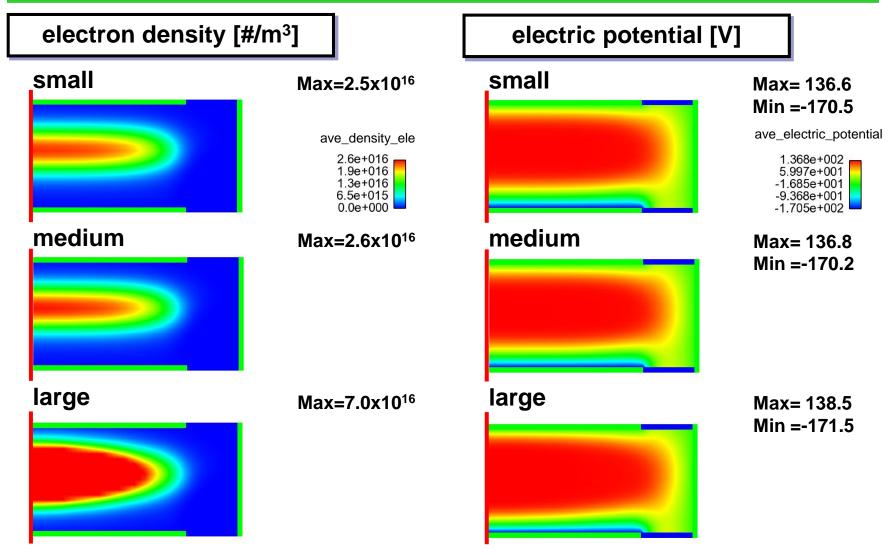
# Mesh size dependence





# Weight of particles dependence



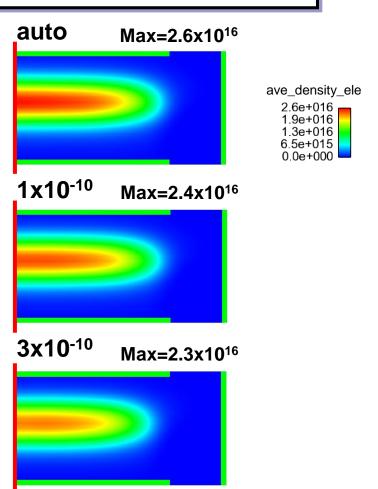


<sup>\*</sup> The weight is the number of molecules represented by a super particle.

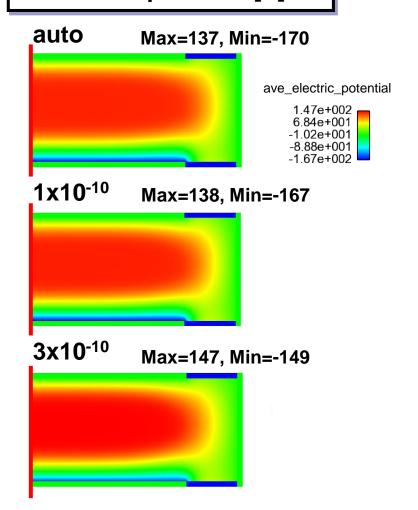
# Time step dependence





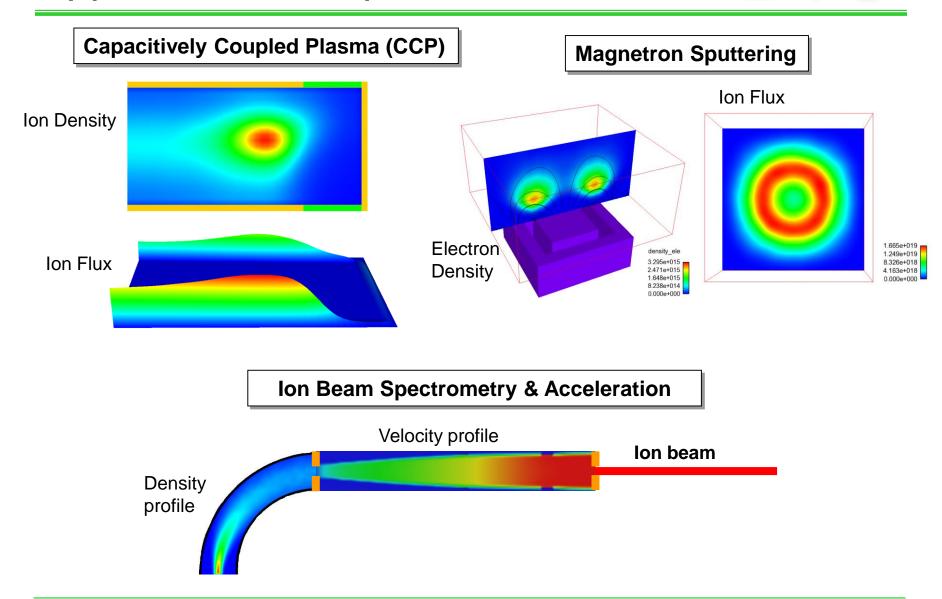


#### electric potential [V]

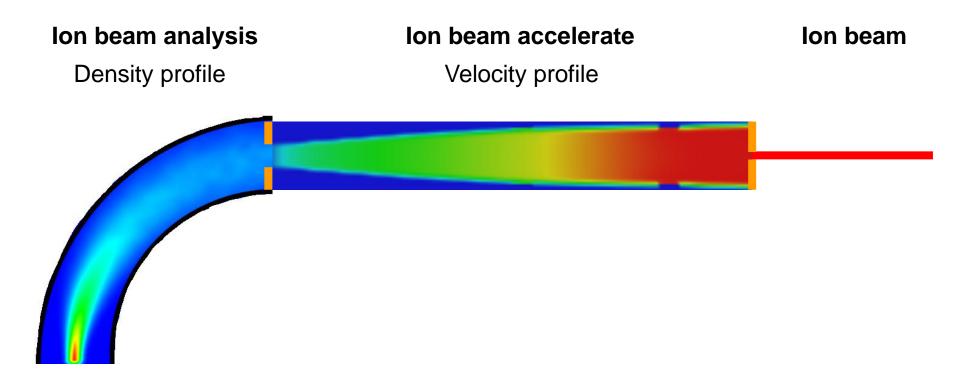


# Application example

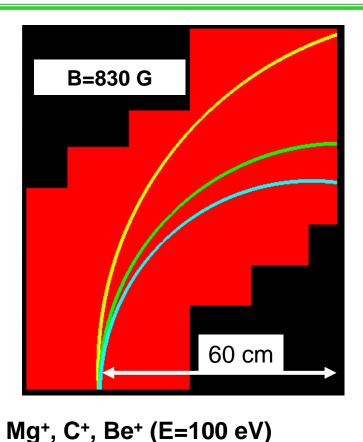












Mg<sup>+</sup> (M=24, R=87 cm)

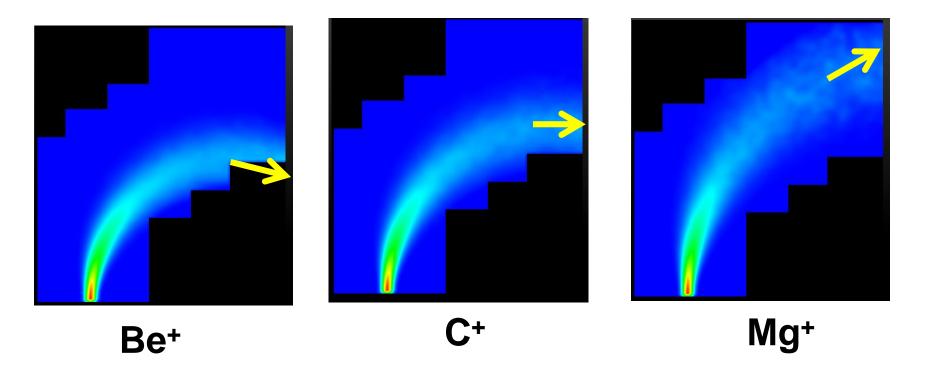
M: molecular weight, R: radius of ion flight pass

#### Radius of ion flight pass

$$R = \frac{mv}{qB} = \frac{\sqrt{2mE}}{qB}$$

The radius of ion flight pass is proportional to root of ion energy. The results of Particle-PLUS agree with analysis results.

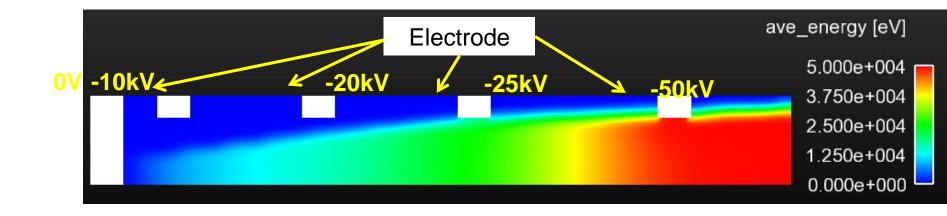




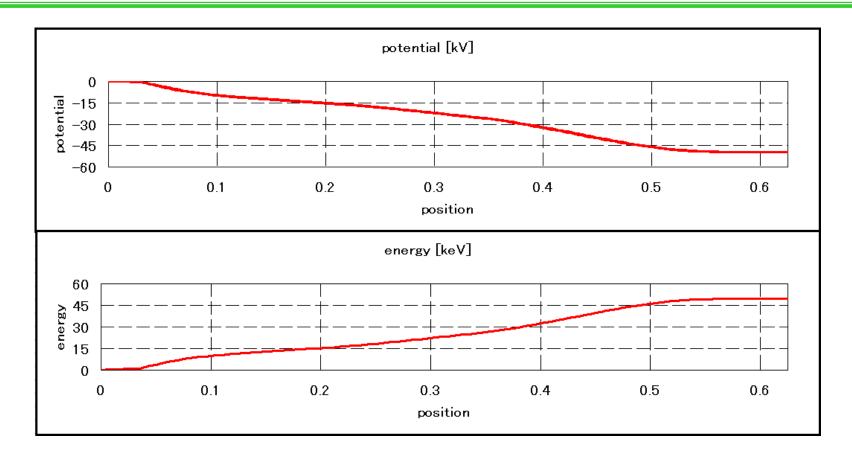


C+: 100 eV

-10 kV -20 kV -25 kV -50 kV



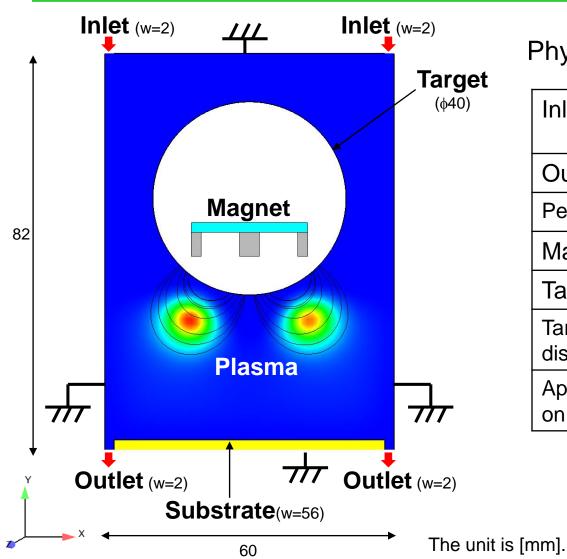




In this simulation, elastic collision is neglected so that sum of potential and kinetic energies is conserved completely.

## 2. Magnetron Sputter Circular target



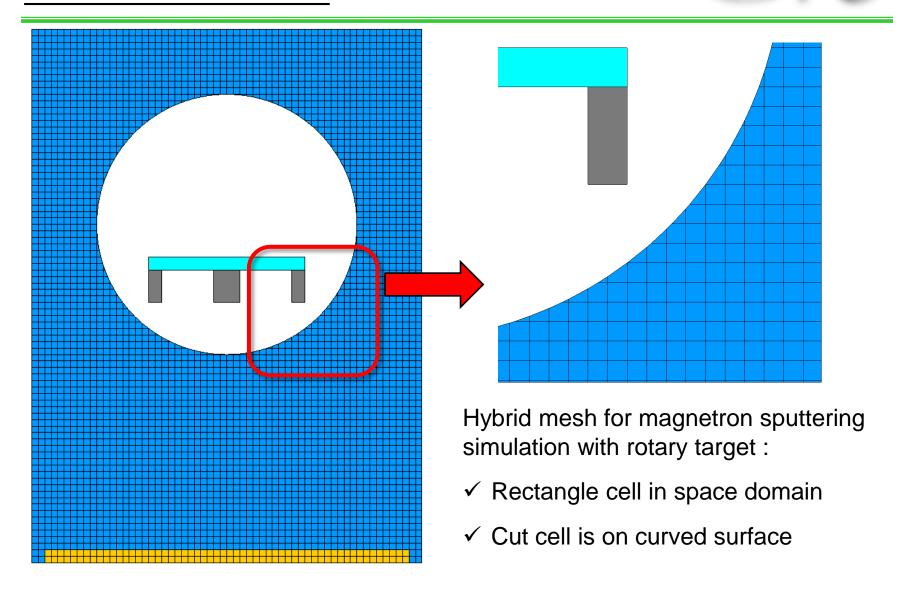


#### Physical Parameters:

Inlet condition	Ar 10 sccm for each part
Outlet condition	1.0 Pa
Permanent magnet	Ferrite
Magnetic Yoke	Fe
Target material	Та
Target-Substrate distance	30 mm
Applied voltage on target	DC -500 V

## 2. Cut cell mesh

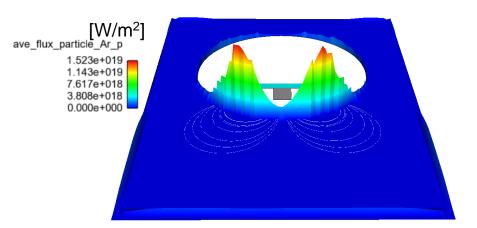




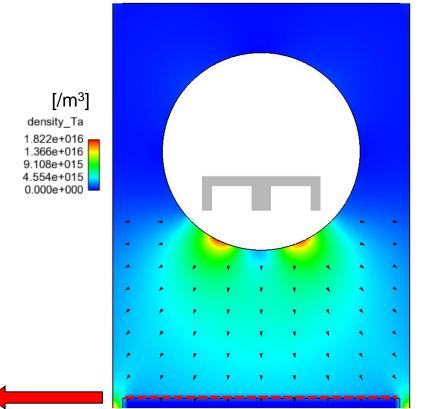
## 2. Results



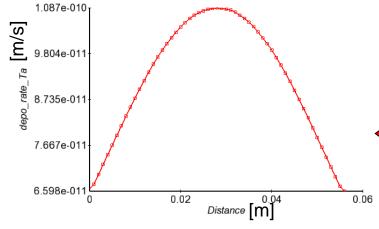
Ar+ Ion Energy Flux (Plasma module)



 Ta Number Density with Flow Vector (Neutral module)



Ta Deposition Rate on Substrate

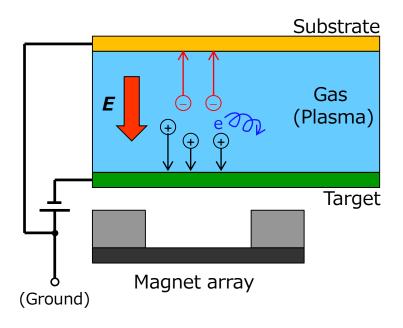


# 3. Magnetron Sputter negative ion



It is easy for halogen gas, for example oxygen and fluorine, to form negative ion because of the large electron affinity. The negative ion damages substrate in DC magnetron sputtering process.

We simulate plasma including negative ions using Particle-PLUS.

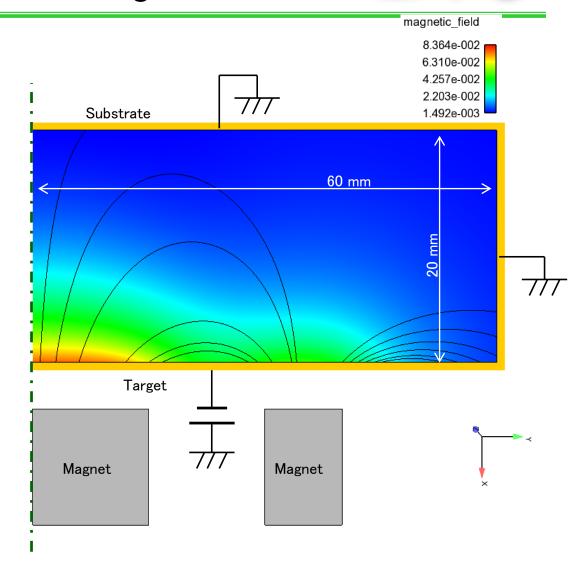


- ✓ Positive ions move according to electric field, and bump into target (sputtering)
- ✓ Negative ions move according to electric field, and bump into substrate (damage)
- ✓ Electrons move with cyclotron motion by magnetic field.

# 3. Magnetron Sputter negative ion



- √ 2D-axisymmetric model
- ✓ <u>Mixture of Ar and O<sub>2</sub>,</u>
  total pressure 1 Pa
  (0.5 Pa + 0.5 Pa)
- ✓ DC -250 [V] on target
- ✓ 2ndary electron emission coefficient 0.02



## 3. Gas Reaction



Collision reaction data set for  $O_2$ -Ar mixture consists of (a) pure  $O_2$  data, (b) pure Ar data, and (c) interactions of  $O_2$  (or products from  $O_2$ ) and Ar. The data set is shown in the following table.

#### • Table a-1. Collision-reactions of pure O<sub>2</sub> system

No.	Collision reaction		Туре	Reference	
(1)	e + O <sub>2</sub>	$\rightarrow$	e + O <sub>2</sub>	Elastic	(Phelps, 1978)
(2)	e + O <sub>2</sub>	$\rightarrow$	e + O <sub>2</sub> *	Excitation	(Phelps, 1978)
(3)	e + O <sub>2</sub>	$\rightarrow$	e + O + O	Dissociation	(Phelps, 1978)
(4)	e + O <sub>2</sub>	$\rightarrow$	e + O + O*	Disso. Excitation	(Phelps, 1978)
(5)	e + O <sub>2</sub>	$\rightarrow$	e + O* + O*	Disso. Excitation	(Phelps, 1978)
(6)	e + O <sub>2</sub>	$\rightarrow$	e + e + O <sub>2</sub> +	Ionization	(Phelps, 1978)
(7)	e + O <sub>2</sub>	$\rightarrow$	O- + O	Disso. Attachment	(Phelps, 1978)
(8)	e + O <sub>2</sub>	$\rightarrow$	$O_2^-$	Attachment	(Phelps, 1978)
(9)	e + O <sub>2</sub>	$\rightarrow$	e + O <sup>+</sup> + O <sup>-</sup>	Ion-pair Formation	(Rapp et al., 1965)
(10)	e + O <sub>2</sub>	$\rightarrow$	e + e + O+ + O	Disso. Ionization	(Eliasson & Kogelschatz, 1986)

## 3. Gas Reaction



#### • Table a-2. Collision-reactions of pure O<sub>2</sub> system

No.	Collision reaction			Туре	Reference
(11)	$O_2^+ + O_2$	$\rightarrow$	$O_2^+ + O_2$	Elastic	(Djilali & Mohammed, 2014)
(12)	$O_2^+ + O_2$	$\rightarrow$	$O_2 + O_2^{+}$	Charge Transfer	(Langevin theory)
(13)	O <sup>-</sup> + O <sub>2</sub>	$\rightarrow$	O <sup>-</sup> + O <sub>2</sub>	Elastic	(Langevin theory)
(14)	O+ + O <sub>2</sub>	$\rightarrow$	O + O <sub>2</sub> +	Charge Transfer	(Langevin theory)

#### • Table b. Collision-reactions of pure Ar system

No.	Collision reacti	ion		Type	Reference
(1)	e + Ar	$\rightarrow$	e + Ar	Elastic	(Yamabe et al., 1983)
(2)	e + Ar	$\rightarrow$	e + Ar*	Excitation	(Yamabe et al., 1983)
(3)	e + Ar	$\rightarrow$	e + e + Ar+	Ionization	(Hayashi, 1987)
(4)	Ar+ + Ar	$\rightarrow$	Ar <sup>+</sup> + Ar	Elastic	(Phelps, 1991)
(5)	Ar+ Ar	$\rightarrow$	Ar + Ar <sup>+</sup>	Charge Transfer	(Phelps, 1991)

## 3. Gas Reaction

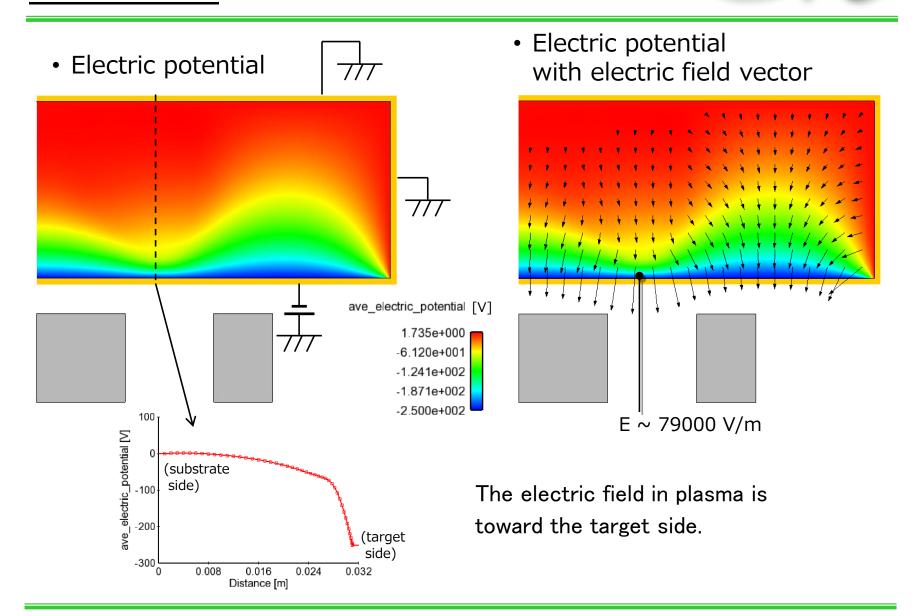


• Table c. additional collision-reactions for O<sub>2</sub>-Ar mixture system

No.	Collision react	ion		Туре	Reference
(1)	O <sub>2</sub> + + Ar	$\rightarrow$	O <sub>2</sub> + + Ar	Elastic	(Langevin theory)
(2)	O <sup>+</sup> + Ar	$\rightarrow$	O+ + Ar	Elastic	(Langevin theory)
(3)	O- + Ar	$\rightarrow$	O- + Ar	Elastic	(Penent et al., 1987)
(4)	$Ar^+ + O_2$	$\rightarrow$	Ar + O <sub>2</sub> +	Charge Transfer	(Langevin theory)

## 3. Results

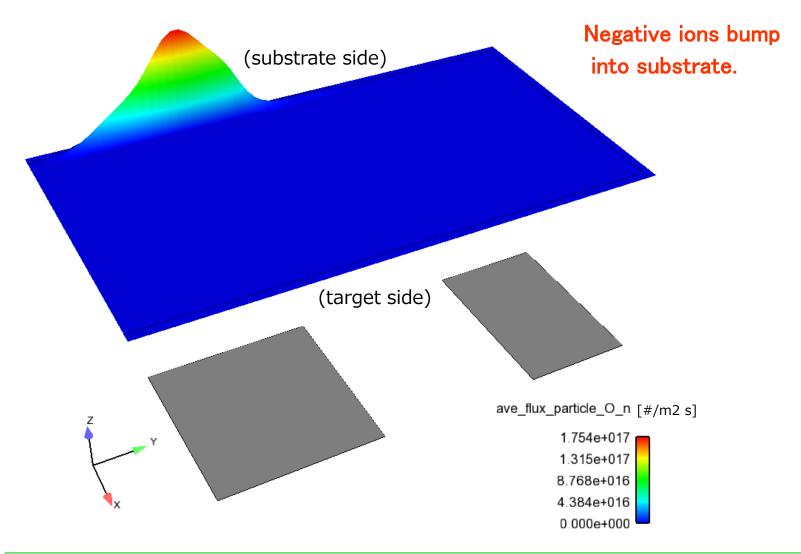




# 3. Negative ion flux



• Number flux of O- ion



# Summary

- DSMC-Neutrals / Particle-PLUS are useful for simulation of magnetron sputtering.
   User can obtain profiles of background gas, magnetic field, electron/ion density, flux, erosion, deposition and so on.
- 2. Particle-PLUS can simulates RF magnetron plasma considering self-bias effect.
- 3. Particle-PLUS can simulates sputtering on circular target.
- 4. Particle-PLUS can simulates 3D magnetron plasma considering cross-corner effect.